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(*E*)-*N*-Benzyl-2-cyano-3-phenylacrylamide

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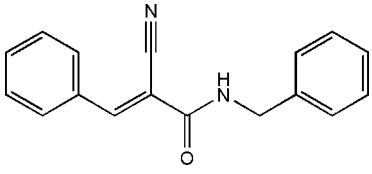
Key indicators: single-crystal X-ray study; T = 291 K; mean σ(C–C) = 0.002 Å; R factor = 0.036; wR factor = 0.096; data-to-parameter ratio = 13.0.

In the title compound, $C_{17}H_{14}N_2O$, the *N*-benzylformamide and phenyl groups are located on the opposite sides of the C=C bond, showing an *E* configuration; the terminal phenyl rings are twisted to each other at a dihedral angle of 63.61 (7)°. Intermolecular classical N—H···N and weak C—H···O hydrogen bonds occur in the crystal structure.

Related literature

For the use of malononitrile-containing compounds as building blocks in syntheses, see: Lee *et al.* (2002); Rajan *et al.* (2001); Yingyongnarongkul *et al.* (2006). For a related structure, see: Kang & Chen (2009).



Experimental

Crystal data

$C_{17}H_{14}N_2O$
M_r = 262.30
Triclinic, $P\overline{1}$
a = 5.8956 (3) Å
b = 9.9224 (5) Å
c = 12.1400 (7) Å
α = 94.508 (5)°
β = 99.544 (4)°
γ = 98.895 (4)°
V = 687.95 (6) Å^3
Z = 2
Cu $K\alpha$ radiation
μ = 0.64 mm^{-1}
T = 291 K
0.36 × 0.35 × 0.30 mm

Data collection

Oxford Diffraction Xcalibur Sapphire3 Gemini ultra diffractometer
Absorption correction: multi-scan (*CrysAlis PRO*; Oxford Diffraction, 2009)
T_{min} = 0.803, T_{max} = 0.832
5416 measured reflections
2407 independent reflections
2202 reflections with $I > 2\sigma(I)$
R_{int} = 0.018

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.036
$wR(F^2)$ = 0.096
S = 1.05
2407 reflections
185 parameters
2 restraints
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.14 e Å^{-3}
$\Delta\rho_{min}$ = −0.12 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N2—H4···N1^{i}	0.88 (1)	2.24 (1)	3.0687 (14)	157 (1)
C3—H3···O1^{ii}	0.93	2.36	3.2672 (16)	164

Symmetry codes: (i) $-x+1, -y, -z+1$; (ii) $-x, -y+1, -z+1$.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2008); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2008); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5098).

References

Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Kang, T.-R. & Chen, L.-M. (2009). *Acta Cryst.* E**65**, o3164.
Lee, S. U., Shin, C. G., Lee, C. K. & Lee, Y. S. (2002). *J. Org. Chem.* **67**, 7019–7028.
Oxford Diffraction (2008). *CrysAlis CCD* and *CrysAlis RED*. Oxford Diffraction Ltd, Yarnton, England.
Oxford Diffraction (2009). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, England.
Rajan, P., Vedernikova, I., Cos, P., Berghe, D. V., Augustyns, K. & Haemers, A. (2001). *Bioorg. Med. Chem. Lett.* **11**, 215–217.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Yingyongnarongkul, B., Apriatikul, N., Aroonrerk, N. & Suksamrarn, A. (2006). *Bioorg. Med. Chem. Lett.* **16**, 5870–5873.

supplementary materials

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(*E*)-*N*-Benzyl-2-cyano-3-phenylacrylamide

T.-R. Kang and L.-M. Chen

Comment

The phenylacrylamide derivatives have broad application for the preparation of heterocyclic ring compounds. The phenylacrylamide derivatives was studied extensively, Rajan (Rajan *et al.*,2001) synthesized a series of phenylacrylamide derivatives and evaluated their antioxidant properties as lipid peroxidation inhibitors. Some phenylacrylamide derivatives and analogues were synthesized and studied antibacterial activity against S. aureus (Yingyongnarongkul *et al.*,2006). Some phenylacrylamide derivatives were synthesized for the purpose of simplifying the structure of *L*-chicoric acid as new HIV-1 integrase inhibitors (Lee *et al.*, 2002).As an extension of this research, we report the synthesis and the crystal structure of the title compound (I), namely, (*E*)—*N*-benzyl-2-cyano-3-phenylacrylamide.

The molecular structure of (I) is shown in Fig. 1. Bond lengths and angles in (I) are normal. The dihedral angle between the C1—C6 and C12—C17 benzene planes is 63.62 (5)°. The crystal packing is stabilized by N—H···N an C—H···O hydrogen bonding (Table 1).

Experimental

N-Benzyl-2-cyanoacetamide (0.258 g, 2 mmol) and benzaldehyde (0.212 g, 2 mmol) were dissolved in 2-propanol (2 ml). To the solution was added piperidine (0.017 g, 0.2 mmol),the solution was stirred for 24 h at 273 K and the solution was filtered to obtain a solid. Recrystallization from hot ethanol afforded the pure compound. Single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation ethanol solvent.

Refinement

Imino H atom was located in a difference Fourier map and refined isotropically. Other H atoms were placed in calculated positions, with C—H = 0.93–0.97 Å, and refined using a riding model, with U_{iso}(H) = 1.2U_{eq}(C).

Figures

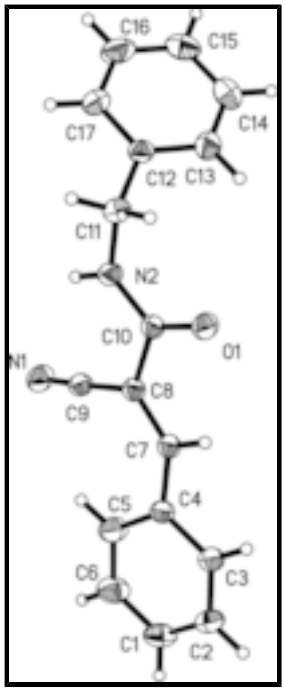

Fig. 1. The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

(*E*)-*N*-Benzyl-2-cyano-3-phenylacrylamide

Crystal data

$C_{17}H_{14}N_2O$	$Z = 2$
$M_r = 262.30$	$F(000) = 276$
Triclinic, $P\overline{1}$	$D_x = 1.266$ Mg m^{-3}
Hall symbol: -P 1	Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å
$a = 5.8956$ (3) Å	Cell parameters from 3956 reflections
$b = 9.9224$ (5) Å	$\theta = 4.5$–$72.2°$
$c = 12.1400$ (7) Å	$\mu = 0.64$ mm^{-1}
$\alpha = 94.508$ (5)°	$T = 291$ K
$\beta = 99.544$ (4)°	Block, yellow
$\gamma = 98.895$ (4)°	0.36 × 0.35 × 0.30 mm
$V = 687.95$ (6) Å^3	

Data collection

Oxford Diffraction Xcalibur Sapphire3 Gemini ultra diffractometer	2407 independent reflections
Radiation source: Enhance Ultra (Cu) X-ray Source	2202 reflections with $I > 2\sigma(I)$
mirror	$R_{int} = 0.018$
Detector resolution: 7.9575 pixels mm^{-1}	$\theta_{max} = 67.1°$, $\theta_{min} = 4.5°$
ω scans	$h = -6 \rightarrow 7$
Absorption correction: multi-scan	$k = -11 \rightarrow 11$

(*CrysAlis PRO*; Oxford Diffraction, 2009)
T_{min} = 0.803, T_{max} = 0.832
5416 measured reflections

l = −14→14

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)]$ = 0.036	Hydrogen site location: inferred from neighbouring sites
$wR(F^2)$ = 0.096	H atoms treated by a mixture of independent and constrained refinement
S = 1.05	$w = 1/[\sigma^2(F_o^2) + (0.0452P)^2 + 0.101P]$ where $P = (F_o^2 + 2F_c^2)/3$
2407 reflections	$(\Delta/\sigma)_{max} < 0.001$
185 parameters	$\Delta\rho_{max}$ = 0.14 e Å^{-3}
2 restraints	$\Delta\rho_{min}$ = −0.12 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	0.28178 (15)	0.43568 (9)	0.56771 (8)	0.0523 (3)
N2	0.53117 (17)	0.28418 (10)	0.58336 (8)	0.0424 (3)
C4	−0.2206 (2)	0.19407 (12)	0.31358 (9)	0.0398 (3)
C12	0.7670 (2)	0.35036 (12)	0.77490 (10)	0.0413 (3)
N1	0.2529 (2)	−0.02233 (11)	0.42829 (10)	0.0556 (3)
C8	0.1591 (2)	0.22182 (11)	0.45635 (9)	0.0371 (3)
C3	−0.3994 (2)	0.26728 (13)	0.27993 (11)	0.0471 (3)
H3	−0.3904	0.3563	0.3128	0.056*
C9	0.20860 (19)	0.08566 (12)	0.43834 (10)	0.0405 (3)
C7	−0.0298 (2)	0.26411 (11)	0.40119 (10)	0.0389 (3)
H7	−0.0409	0.3545	0.4229	0.047*
C11	0.7242 (2)	0.37901 (13)	0.65408 (11)	0.0458 (3)
H11A	0.6918	0.4717	0.6509	0.055*
H11B	0.8653	0.3745	0.6237	0.055*

C10	0.32978 (19)	0.32315 (11)	0.54175 (9)	0.0375 (3)
C2	−0.5896 (2)	0.20970 (15)	0.19856 (12)	0.0570 (4)
H2	−0.7080	0.2596	0.1776	0.068*
C17	0.9529 (2)	0.28819 (16)	0.81647 (12)	0.0576 (4)
H17	1.0518	0.2634	0.7691	0.069*
C15	0.8499 (3)	0.29793 (18)	0.99859 (13)	0.0689 (4)
H15	0.8778	0.2804	1.0733	0.083*
C13	0.6229 (2)	0.38575 (14)	0.84780 (12)	0.0536 (3)
H13	0.4967	0.4276	0.8216	0.064*
C5	−0.2386 (3)	0.06160 (14)	0.26135 (12)	0.0579 (4)
H5	−0.1209	0.0109	0.2813	0.069*
C6	−0.4295 (3)	0.00533 (15)	0.18032 (13)	0.0666 (4)
H6	−0.4404	−0.0835	0.1467	0.080*
C14	0.6643 (3)	0.35969 (16)	0.95881 (13)	0.0646 (4)
H14	0.5661	0.3840	1.0067	0.078*
C1	−0.6041 (3)	0.07908 (15)	0.14864 (12)	0.0613 (4)
H1	−0.7318	0.0405	0.0935	0.074*
C16	0.9936 (3)	0.26243 (19)	0.92760 (14)	0.0727 (5)
H16	1.1195	0.2206	0.9543	0.087*
H4	0.556 (2)	0.2006 (13)	0.5647 (12)	0.054 (4)*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0492 (5)	0.0412 (5)	0.0618 (6)	0.0158 (4)	−0.0034 (4)	−0.0101 (4)
N2	0.0437 (6)	0.0402 (5)	0.0415 (6)	0.0134 (4)	−0.0015 (4)	−0.0002 (4)
C4	0.0431 (6)	0.0387 (6)	0.0364 (6)	0.0082 (5)	0.0028 (5)	0.0033 (5)
C12	0.0383 (6)	0.0394 (6)	0.0423 (6)	0.0043 (5)	0.0002 (5)	0.0003 (5)
N1	0.0566 (7)	0.0418 (6)	0.0664 (7)	0.0175 (5)	−0.0001 (5)	0.0008 (5)
C8	0.0409 (6)	0.0347 (5)	0.0360 (6)	0.0089 (4)	0.0059 (5)	0.0023 (4)
C3	0.0502 (7)	0.0461 (7)	0.0434 (7)	0.0138 (5)	0.0008 (5)	0.0012 (5)
C9	0.0406 (6)	0.0386 (6)	0.0406 (6)	0.0097 (5)	0.0006 (5)	0.0016 (5)
C7	0.0436 (6)	0.0345 (6)	0.0388 (6)	0.0098 (5)	0.0061 (5)	0.0016 (5)
C11	0.0398 (6)	0.0491 (7)	0.0459 (7)	0.0060 (5)	0.0021 (5)	0.0051 (5)
C10	0.0405 (6)	0.0370 (6)	0.0355 (6)	0.0100 (5)	0.0059 (5)	0.0033 (4)
C2	0.0504 (8)	0.0660 (9)	0.0508 (8)	0.0160 (6)	−0.0066 (6)	0.0042 (6)
C17	0.0460 (7)	0.0747 (9)	0.0524 (8)	0.0190 (6)	0.0018 (6)	0.0059 (7)
C15	0.0760 (10)	0.0812 (11)	0.0422 (8)	0.0034 (8)	−0.0028 (7)	0.0101 (7)
C13	0.0560 (8)	0.0536 (8)	0.0543 (8)	0.0179 (6)	0.0109 (6)	0.0049 (6)
C5	0.0632 (8)	0.0441 (7)	0.0597 (8)	0.0163 (6)	−0.0101 (7)	−0.0042 (6)
C6	0.0784 (10)	0.0460 (8)	0.0625 (9)	0.0067 (7)	−0.0133 (8)	−0.0094 (7)
C14	0.0777 (10)	0.0672 (9)	0.0505 (8)	0.0113 (8)	0.0197 (7)	0.0003 (7)
C1	0.0586 (8)	0.0619 (9)	0.0514 (8)	−0.0007 (7)	−0.0119 (6)	0.0010 (7)
C16	0.0598 (9)	0.0964 (13)	0.0599 (9)	0.0238 (8)	−0.0093 (7)	0.0186 (9)

Geometric parameters (Å, °)

O1—C10	1.2240 (13)	C11—H11B	0.9700
N2—C10	1.3372 (14)	C2—C1	1.371 (2)

N2—C11	1.4612 (15)	C2—H2	0.9300
N2—H4	0.884 (12)	C17—C16	1.382 (2)
C4—C5	1.3945 (17)	C17—H17	0.9300
C4—C3	1.3954 (16)	C15—C16	1.368 (2)
C4—C7	1.4574 (16)	C15—C14	1.373 (2)
C12—C17	1.3824 (17)	C15—H15	0.9300
C12—C13	1.3862 (18)	C13—C14	1.381 (2)
C12—C11	1.5036 (17)	C13—H13	0.9300
N1—C9	1.1435 (15)	C5—C6	1.3781 (19)
C8—C7	1.3443 (16)	C5—H5	0.9300
C8—C9	1.4334 (15)	C6—C1	1.374 (2)
C8—C10	1.5084 (16)	C6—H6	0.9300
C3—C2	1.3803 (18)	C14—H14	0.9300
C3—H3	0.9300	C1—H1	0.9300
C7—H7	0.9300	C16—H16	0.9300
C11—H11A	0.9700		
C10—N2—C11	122.18 (10)	C1—C2—C3	120.08 (13)
C10—N2—H4	120.6 (9)	C1—C2—H2	120.0
C11—N2—H4	117.1 (9)	C3—C2—H2	120.0
C5—C4—C3	117.89 (11)	C16—C17—C12	120.79 (14)
C5—C4—C7	125.69 (11)	C16—C17—H17	119.6
C3—C4—C7	116.41 (10)	C12—C17—H17	119.6
C17—C12—C13	118.10 (12)	C16—C15—C14	119.57 (14)
C17—C12—C11	120.65 (12)	C16—C15—H15	120.2
C13—C12—C11	121.25 (11)	C14—C15—H15	120.2
C7—C8—C9	123.69 (10)	C14—C13—C12	120.87 (13)
C7—C8—C10	118.37 (10)	C14—C13—H13	119.6
C9—C8—C10	117.94 (9)	C12—C13—H13	119.6
C2—C3—C4	121.03 (12)	C6—C5—C4	120.46 (12)
C2—C3—H3	119.5	C6—C5—H5	119.8
C4—C3—H3	119.5	C4—C5—H5	119.8
N1—C9—C8	177.26 (13)	C1—C6—C5	120.71 (13)
C8—C7—C4	131.72 (10)	C1—C6—H6	119.6
C8—C7—H7	114.1	C5—C6—H6	119.6
C4—C7—H7	114.1	C15—C14—C13	120.19 (14)
N2—C11—C12	113.92 (10)	C15—C14—H14	119.9
N2—C11—H11A	108.8	C13—C14—H14	119.9
C12—C11—H11A	108.8	C2—C1—C6	119.82 (13)
N2—C11—H11B	108.8	C2—C1—H1	120.1
C12—C11—H11B	108.8	C6—C1—H1	120.1
H11A—C11—H11B	107.7	C15—C16—C17	120.48 (14)
O1—C10—N2	123.51 (11)	C15—C16—H16	119.8
O1—C10—C8	119.71 (10)	C17—C16—H16	119.8
N2—C10—C8	116.76 (10)		
C5—C4—C3—C2	0.8 (2)	C9—C8—C10—N2	8.94 (16)
C7—C4—C3—C2	−179.17 (12)	C4—C3—C2—C1	−0.6 (2)
C7—C8—C9—N1	−153 (3)	C13—C12—C17—C16	−0.1 (2)
C10—C8—C9—N1	27 (3)	C11—C12—C17—C16	179.64 (13)

C9—C8—C7—C4	−1.4 (2)	C17—C12—C13—C14	0.1 (2)
C10—C8—C7—C4	178.30 (11)	C11—C12—C13—C14	−179.67 (12)
C5—C4—C7—C8	−4.2 (2)	C3—C4—C5—C6	−0.9 (2)
C3—C4—C7—C8	175.79 (12)	C7—C4—C5—C6	179.10 (14)
C10—N2—C11—C12	109.92 (13)	C4—C5—C6—C1	0.7 (3)
C17—C12—C11—N2	104.01 (14)	C16—C15—C14—C13	0.0 (3)
C13—C12—C11—N2	−76.26 (15)	C12—C13—C14—C15	0.0 (2)
C11—N2—C10—O1	−7.34 (18)	C3—C2—C1—C6	0.4 (2)
C11—N2—C10—C8	171.11 (10)	C5—C6—C1—C2	−0.5 (3)
C7—C8—C10—O1	7.69 (17)	C14—C15—C16—C17	0.0 (3)
C9—C8—C10—O1	−172.55 (11)	C12—C17—C16—C15	0.1 (3)
C7—C8—C10—N2	−170.82 (11)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N2—H4···N1^{i}	0.88 (1)	2.24 (1)	3.0687 (14)	157 (1)
C3—H3···O1^{ii}	0.93	2.36	3.2672 (16)	164

Symmetry codes: (i) −*x*+1, −*y*, −*z*+1; (ii) −*x*, −*y*+1, −*z*+1.

Fig. 1